

# Benzamide, 2-fluoro-N-decyl-

<b>Inchi:</b>	InChI=1S/C17H26FNO/c1-2-3-4-5-6-7-8-11-14-19-17(20)15-12-9-10-13-16(15)18/h9-10,
<b>InchiKey:</b>	SKMAAOXQTGGZJX-UHFFFAOYSA-N
<b>Formula:</b>	C17H26FNO
<b>SMILES:</b>	CCCCCCCCCNC(=O)c1ccccc1F
<b>Mol. weight [g/mol]:</b>	279.39

## Physical Properties

Property code	Value	Unit	Source
gf	-39.30	kJ/mol	Joback Method
hf	-424.37	kJ/mol	Joback Method
hfus	43.22	kJ/mol	Joback Method
hvap	68.74	kJ/mol	Joback Method
log10ws	-5.92		Crippen Method
logp	4.696		Crippen Method
mvol	239.950	ml/mol	McGowan Method
pc	1596.17	kPa	Joback Method
rinpol	2222.00		NIST Webbook
rinpol	2222.00		NIST Webbook
tb	723.33	K	Joback Method
tc	913.59	K	Joback Method
tf	423.47	K	Joback Method
vc	0.939	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	692.64	J/mol×K	723.33	Joback Method
cpg	709.09	J/mol×K	755.04	Joback Method
cpg	724.61	J/mol×K	786.75	Joback Method
cpg	739.25	J/mol×K	818.46	Joback Method
cpg	753.03	J/mol×K	850.17	Joback Method
cpg	766.00	J/mol×K	881.88	Joback Method
cpg	778.19	J/mol×K	913.59	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U407140&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U407140&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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