

Phthalic acid, 3,5-dinitro-4-methylbenzyl ethyl ester

Inchi:	InChI=1S/C18H16N2O8/c1-3-27-17(21)13-6-4-5-7-14(13)18(22)28-10-12-8-15(19(23)24)
InchiKey:	UMCIBIMLRJDHCU-UHFFFAOYSA-N
Formula:	C18H16N2O8
SMILES:	CCOC(=O)c1cccc1C(=O)OCc1cc([N+](=O)[O-])c(C)c([N+](=O)[O-])c1
Mol. weight [g/mol]:	388.33

Physical Properties

Property code	Value	Unit	Source
gf	-109.76	kJ/mol	Joback Method
hf	-498.79	kJ/mol	Joback Method
hfus	57.20	kJ/mol	Joback Method
hvap	114.36	kJ/mol	Joback Method
log10ws	-6.26		Crippen Method
logp	3.345		Crippen Method
mvol	266.680	ml/mol	McGowan Method
pc	2014.51	kPa	Joback Method
rinpol	3027.00		NIST Webbook
rinpol	3027.00		NIST Webbook
tb	1140.78	K	Joback Method
tc	1408.95	K	Joback Method
tf	827.08	K	Joback Method
vc	1.040	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	831.24	J/mol×K	1140.78	Joback Method
cpg	835.63	J/mol×K	1185.48	Joback Method
cpg	838.39	J/mol×K	1230.17	Joback Method
cpg	839.54	J/mol×K	1274.87	Joback Method
cpg	839.11	J/mol×K	1319.56	Joback Method
cpg	837.15	J/mol×K	1364.26	Joback Method
cpg	833.69	J/mol×K	1408.95	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U415509&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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