

# Pentafluorobenzoic acid, 2-propylphenyl ester

<b>Other names:</b>	2-n-Propylphenol, pentafluorobenzoyl ester
<b>Inchi:</b>	InChI=1S/C16H11F5O2/c1-2-5-8-6-3-4-7-9(8)23-16(22)10-11(17)13(19)15(21)14(20)12(16)
<b>InchiKey:</b>	IEJFGMQLJBOJQU-UHFFFAOYSA-N
<b>Formula:</b>	C16H11F5O2
<b>SMILES:</b>	CCCc1ccccc1OC(=O)c1c(F)c(F)c(F)c(F)c1F
<b>Mol. weight [g/mol]:</b>	330.25

## Physical Properties

Property code	Value	Unit	Source
gf	-957.09	kJ/mol	Joback Method
hf	-1194.68	kJ/mol	Joback Method
hfus	41.13	kJ/mol	Joback Method
hvap	64.81	kJ/mol	Joback Method
log10ws	-6.44		Crippen Method
logp	4.554		Crippen Method
mcvol	205.070	ml/mol	McGowan Method
pc	1793.94	kPa	Joback Method
rinpol	1688.00		NIST Webbook
rinpol	1685.90		NIST Webbook
rinpol	1707.00		NIST Webbook
rinpol	1690.70		NIST Webbook
rinpol	1707.00		NIST Webbook
rinpol	1685.90		NIST Webbook
tb	721.36	K	Joback Method
tc	916.78	K	Joback Method
tf	473.15	K	Joback Method
vc	0.830	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	552.28	J/molxK	721.36	Joback Method
cpg	564.43	J/molxK	753.93	Joback Method
cpg	575.82	J/molxK	786.50	Joback Method

cpg	586.46	J/mol×K	819.07	Joback Method
cpg	596.36	J/mol×K	851.64	Joback Method
cpg	605.53	J/mol×K	884.21	Joback Method
cpg	613.98	J/mol×K	916.78	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U360627&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U360627&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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