

# L-Valine, N-(3-fluoro-4-trifluoromethylbenzoyl)-, butyl

Inchi:  
ester

InChI=1S/C17H21F4NO3/c1-4-5-8-25-16(24)14(10(2)3)22-15(23)11-6-7-12(13(18)9-11)1

InchiKey:

HUOOHLBMNVKAGK-UHFFFAOYSA-N

Formula:

C17H21F4NO3

SMILES:

CCCCOC(=O)C(NC(=O)c1ccc(C(F)(F)F)c(F)c1)C(C)C

Mol. weight [g/mol]:

363.35

## Physical Properties

Property code	Value	Unit	Source
gf	-869.32	kJ/mol	Joback Method
hf	-1288.28	kJ/mol	Joback Method
hfus	40.39	kJ/mol	Joback Method
hvap	74.03	kJ/mol	Joback Method
log10ws	-5.33		Crippen Method
logp	3.942		Crippen Method
mcvol	252.700	ml/mol	McGowan Method
pc	1511.67	kPa	Joback Method
rinpol	1905.00		NIST Webbook
rinpol	1905.00		NIST Webbook
tb	798.30	K	Joback Method
tc	991.98	K	Joback Method
tf	482.34	K	Joback Method
vc	0.994	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	765.41	J/mol×K	798.30	Joback Method
cpg	779.02	J/mol×K	830.58	Joback Method
cpg	791.69	J/mol×K	862.86	Joback Method
cpg	803.47	J/mol×K	895.14	Joback Method
cpg	814.39	J/mol×K	927.42	Joback Method
cpg	824.50	J/mol×K	959.70	Joback Method
cpg	833.82	J/mol×K	991.98	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U346552&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U346552&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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