

# I-Valine, N-pivaloyl-, methyl ester

<b>Inchi:</b>	InChI=1S/C11H21NO3/c1-7(2)8(9(13)15-6)12-10(14)11(3,4)5/h7-8H,1-6H3,(H,12,14)
<b>InchiKey:</b>	ZOMUUMHLEGAWIX-UHFFFAOYSA-N
<b>Formula:</b>	C11H21NO3
<b>SMILES:</b>	COC(=O)C(NC(=O)C(C)(C)C)C(C)C
<b>Mol. weight [g/mol]:</b>	215.29

## Physical Properties

Property code	Value	Unit	Source
gf	-233.75	kJ/mol	Joback Method
hf	-593.59	kJ/mol	Joback Method
hfus	19.27	kJ/mol	Joback Method
hvap	60.35	kJ/mol	Joback Method
log10ws	-1.88		Crippen Method
logp	1.346		Crippen Method
mcvol	184.840	ml/mol	McGowan Method
pc	2231.30	kPa	Joback Method
rinpol	1343.00		NIST Webbook
rinpol	1343.00		NIST Webbook
tb	627.30	K	Joback Method
tc	824.24	K	Joback Method
tf	360.90	K	Joback Method
vc	0.694	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	496.08	J/mol×K	627.30	Joback Method
cpg	511.25	J/mol×K	660.12	Joback Method
cpg	525.57	J/mol×K	692.95	Joback Method
cpg	539.05	J/mol×K	725.77	Joback Method
cpg	551.72	J/mol×K	758.59	Joback Method
cpg	563.62	J/mol×K	791.42	Joback Method
cpg	574.76	J/mol×K	824.24	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U299743&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U299743&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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