

Succinic acid, 2,2,3,3,4,4,4-heptafluorobutyl 2-methylhex-3-yl ester

Inchi:	InChI=1S/C15H21F7O4/c1-4-5-10(9(2)3)26-12(24)7-6-11(23)25-8-13(16,17)14(18,19)15
InchiKey:	IAWYKDRFTPIBSU-UHFFFAOYSA-N
Formula:	C15H21F7O4
SMILES:	CCCC(OC(=O)CCC(=O)OCC(F)(F)C(F)(F)C(F)(F)C(F)(F)C(C)C
Mol. weight [g/mol]:	398.31

Physical Properties

Property code	Value	Unit	Source
gf	-1752.45	kJ/mol	Joback Method
hf	-2252.11	kJ/mol	Joback Method
hfus	32.45	kJ/mol	Joback Method
hvap	56.91	kJ/mol	Joback Method
log10ws	-4.99		Crippen Method
logp	4.511		Crippen Method
mvol	249.480	ml/mol	McGowan Method
pc	1268.25	kPa	Joback Method
rmpol	1461.00		NIST Webbook
rmpol	1461.00		NIST Webbook
tb	679.50	K	Joback Method
tc	843.39	K	Joback Method
tf	384.52	K	Joback Method
vc	1.004	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	740.54	J/molxK	679.50	Joback Method
cpg	754.75	J/molxK	706.82	Joback Method
cpg	768.14	J/molxK	734.13	Joback Method
cpg	780.75	J/molxK	761.45	Joback Method
cpg	792.62	J/molxK	788.76	Joback Method
cpg	803.78	J/molxK	816.08	Joback Method
cpg	814.27	J/molxK	843.39	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U382352&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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