

# Glutaric acid, cyclohexylmethyl 2,6-dichlorophenyl ester

**Inchi:** InChI=1S/C18H22Cl2O4/c19-14-8-4-9-15(20)18(14)24-17(22)11-5-10-16(21)23-12-13-6-  
**InchiKey:** QGOQWBKEEXHMHE-UHFFFAOYSA-N  
**Formula:** C18H22Cl2O4  
**SMILES:** O=C(CCCC(=O)Oc1c(Cl)cccc1Cl)OCC1CCCCC1  
**Mol. weight [g/mol]:** 373.27

## Physical Properties

Property code	Value	Unit	Source
gf	-273.42	kJ/mol	Joback Method
hf	-668.02	kJ/mol	Joback Method
hfus	41.44	kJ/mol	Joback Method
hvap	86.77	kJ/mol	Joback Method
log10ws	-5.86		Crippen Method
logp	5.193		Crippen Method
mvol	269.220	ml/mol	McGowan Method
pc	1704.71	kPa	Joback Method
rinpol	2711.00		NIST Webbook
rinpol	2711.00		NIST Webbook
tb	894.87	K	Joback Method
tc	1125.18	K	Joback Method
tf	555.62	K	Joback Method
vc	1.014	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	805.69	J/molxK	894.87	Joback Method
cpg	819.49	J/molxK	933.25	Joback Method
cpg	831.82	J/molxK	971.64	Joback Method
cpg	842.72	J/molxK	1010.02	Joback Method
cpg	852.22	J/molxK	1048.41	Joback Method
cpg	860.34	J/molxK	1086.79	Joback Method
cpg	867.11	J/molxK	1125.18	Joback Method
dvisc	0.0004838	Paxs	555.62	Joback Method

dvisc	0.0002871	Paxs	612.16	Joback Method
dvisc	0.0001861	Paxs	668.70	Joback Method
dvisc	0.0001290	Paxs	725.24	Joback Method
dvisc	0.0000943	Paxs	781.79	Joback Method
dvisc	0.0000720	Paxs	838.33	Joback Method
dvisc	0.0000568	Paxs	894.87	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U390260&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U390260&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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