

2,2',4,4'-Tetramethyldiphenylsulphone

Other names:	Benzene, 1,1'-sulfonylbis[2,4-dimethyl-di-2,4-xylyl sulphone
Inchi:	InChI=1S/C16H18O2S/c1-11-5-7-15(13(3)9-11)19(17,18)16-8-6-12(2)10-14(16)4/h5-10H
InchiKey:	CWZZFEROKOHBGT-UHFFFAOYSA-N
Formula:	C16H18O2S
SMILES:	<chem>Cc1ccc(S(=O)(=O)c2ccc(C)cc2C)c(C)c1</chem>
Mol. weight [g/mol]:	274.38
CAS:	5184-75-8

Physical Properties

Property code	Value	Unit	Source
gf	-198.40	kJ/mol	Joback Method
hf	-399.74	kJ/mol	Joback Method
hfus	35.10	kJ/mol	Joback Method
hvap	77.05	kJ/mol	Joback Method
log10ws	-4.74		Crippen Method
logp	3.753		Crippen Method
mcvol	216.870	ml/mol	McGowan Method
pc	2507.52	kPa	Joback Method
tb	686.54	K	Joback Method
tc	910.52	K	Joback Method
tf	393.00 ± 2.00	K	NIST Webbook
vc	0.842	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	559.18	J/molxK	686.54	Joback Method
cpg	576.01	J/molxK	723.87	Joback Method
cpg	591.65	J/molxK	761.20	Joback Method
cpg	606.12	J/molxK	798.53	Joback Method
cpg	619.43	J/molxK	835.86	Joback Method
cpg	631.60	J/molxK	873.19	Joback Method
cpg	642.67	J/molxK	910.52	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C5184758&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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