

# Trichloroacetic acid, undecyl ester

<b>Other names:</b>	Undecyl trichloroacetate
<b>Inchi:</b>	InChI=1S/C13H23Cl3O2/c1-2-3-4-5-6-7-8-9-10-11-18-12(17)13(14,15)16/h2-11H2,1H3
<b>InchiKey:</b>	ZKEYPXQCWNXAQG-UHFFFAOYSA-N
<b>Formula:</b>	C13H23Cl3O2
<b>SMILES:</b>	CCCCCCCCCCCCOC(=O)C(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	317.68
<b>CAS:</b>	74339-49-4

## Physical Properties

Property code	Value	Unit	Source
gf	-208.29	kJ/mol	Joback Method
hf	-612.42	kJ/mol	Joback Method
hfus	37.39	kJ/mol	Joback Method
hvap	65.55	kJ/mol	Joback Method
log10ws	-5.69		Crippen Method
logp	5.431		Crippen Method
mcvol	238.190	ml/mol	McGowan Method
pc	1584.75	kPa	Joback Method
rinpol	1864.00		NIST Webbook
rinpol	1864.00		NIST Webbook
rinpol	1871.00		NIST Webbook
rinpol	1871.00		NIST Webbook
ripol	2263.00		NIST Webbook
ripol	2263.00		NIST Webbook
tb	682.19	K	Joback Method
tc	872.30	K	Joback Method
tf	400.61	K	Joback Method
vc	0.923	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	611.33	J/molxK	682.19	Joback Method
cpg	625.80	J/molxK	713.87	Joback Method

cpg	639.46	J/molxK	745.56	Joback Method
cpg	652.33	J/molxK	777.24	Joback Method
cpg	664.47	J/molxK	808.93	Joback Method
cpg	675.89	J/molxK	840.61	Joback Method
cpg	686.64	J/molxK	872.30	Joback Method
dvisc	0.0016819	Paxs	400.61	Joback Method
dvisc	0.0008389	Paxs	447.54	Joback Method
dvisc	0.0004775	Paxs	494.47	Joback Method
dvisc	0.0002997	Paxs	541.40	Joback Method
dvisc	0.0002026	Paxs	588.33	Joback Method
dvisc	0.0001451	Paxs	635.26	Joback Method
dvisc	0.0001088	Paxs	682.19	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C74339494&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C74339494&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/114-410-9/Trichloroacetic-acid-undecyl-ester.pdf>

Generated by Cheméo on 2024-04-29 06:04:35.889017763 +0000 UTC m=+16659924.809595075.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.