

Succinic acid, 2-chloro-6-fluorophenyl 2-fluoro-3-(trifluoromethyl)phenyl ester

Inchi:	InChI=1S/C17H10ClF5O4/c18-10-4-2-5-11(19)16(10)27-14(25)8-7-13(24)26-12-6-1-3-9(
InchiKey:	XQEWRVKXKJTJDD-UHFFFAOYSA-N
Formula:	C17H10ClF5O4
SMILES:	O=C(CCC(=O)Oc1c(F)cccc1Cl)Oc1cccc(C(F)(F)F)c1F
Mol. weight [g/mol]:	408.70

Physical Properties

Property code	Value	Unit	Source
gf	-1172.42	kJ/mol	Joback Method
hf	-1461.67	kJ/mol	Joback Method
hfus	44.07	kJ/mol	Joback Method
hvap	77.95	kJ/mol	Joback Method
log10ws	-6.20		Crippen Method
logp	4.928		Crippen Method
mcvol	238.840	ml/mol	McGowan Method
pc	1737.56	kPa	Joback Method
rinpol	2256.00		NIST Webbook
rinpol	2256.00		NIST Webbook
tb	844.77	K	Joback Method
tc	1055.20	K	Joback Method
tf	563.88	K	Joback Method
vc	0.948	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	670.21	J/mol×K	844.77	Joback Method
cpg	680.16	J/mol×K	879.84	Joback Method
cpg	689.17	J/mol×K	914.91	Joback Method
cpg	697.26	J/mol×K	949.99	Joback Method
cpg	704.48	J/mol×K	985.06	Joback Method
cpg	710.85	J/mol×K	1020.13	Joback Method
cpg	716.40	J/mol×K	1055.20	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U390787&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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