

Hexanoic acid, 3,5,5-trimethyl-, 2-chloro-6-fluorophenyl ester

Inchi:	InChI=1S/C15H20ClFO2/c1-10(9-15(2,3)4)8-13(18)19-14-11(16)6-5-7-12(14)17/h5-7,10H
InchiKey:	XPLONZOJIRMXHQ-UHFFFAOYSA-N
Formula:	C15H20ClFO2
SMILES:	CC(CC(=O)Oc1c(F)cccc1Cl)CC(C)(C)C
Mol. weight [g/mol]:	286.77

Physical Properties

Property code	Value	Unit	Source
gf	-271.69	kJ/mol	Joback Method
hf	-610.02	kJ/mol	Joback Method
hfus	27.00	kJ/mol	Joback Method
hvap	63.62	kJ/mol	Joback Method
log10ws	-5.25		Crippen Method
logp	4.847		Crippen Method
mvol	219.900	ml/mol	McGowan Method
pc	1786.37	kPa	Joback Method
rinpol	1743.00		NIST Webbook
rinpol	1743.00		NIST Webbook
tb	688.56	K	Joback Method
tc	896.76	K	Joback Method
tf	400.36	K	Joback Method
vc	0.842	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	586.11	J/mol×K	688.56	Joback Method
cpg	601.65	J/mol×K	723.26	Joback Method
cpg	616.20	J/mol×K	757.96	Joback Method
cpg	629.79	J/mol×K	792.66	Joback Method
cpg	642.47	J/mol×K	827.36	Joback Method
cpg	654.29	J/mol×K	862.06	Joback Method
cpg	665.27	J/mol×K	896.76	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U406824&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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