

Tiapride

Inchi:	InChI=1S/C15H24N2O4S/c1-5-17(6-2)10-9-16-15(18)13-11-12(22(4,19)20)7-8-14(13)21
InchiKey:	JTVPZMFULRWINT-UHFFFAOYSA-N
Formula:	C15H24N2O4S
SMILES:	CCN(CC)CCNC(=O)c1cc(S(C)(=O)=O)ccc1OC
Mol. weight [g/mol]:	328.43
CAS:	51012-32-9

Physical Properties

Property code	Value	Unit	Source
gf	-333.72	kJ/mol	Joback Method
hf	-716.49	kJ/mol	Joback Method
hfus	50.15	kJ/mol	Joback Method
hvap	88.85	kJ/mol	Joback Method
log10ws	-2.29		Crippen Method
logp	1.170		Crippen Method
mcvol	253.940	ml/mol	McGowan Method
pc	2173.43	kPa	Joback Method
rinpol	2850.00		NIST Webbook
rinpol	2850.00		NIST Webbook
tb	765.92	K	Joback Method
tc	961.20	K	Joback Method
tf	506.12	K	Joback Method
vc	0.971	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	734.43	J/molxK	765.92	Joback Method
cpg	749.59	J/molxK	798.47	Joback Method
cpg	763.68	J/molxK	831.01	Joback Method
cpg	776.72	J/molxK	863.56	Joback Method
cpg	788.72	J/molxK	896.10	Joback Method
cpg	799.70	J/molxK	928.65	Joback Method
cpg	809.67	J/molxK	961.20	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C51012329&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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