

# Sebacic acid, dodecyl 2,2,3,3,4,4,5,5-octafluoropentyl ester

Inchi: InChI=1S/C27H44F8O4/c1-2-3-4-5-6-7-8-11-14-17-20-38-22(36)18-15-12-9-10-13-16-19

InchiKey: URRVNUBITQMNRX-UHFFFAOYSA-N

Formula: C27H44F8O4

SMILES: CCCCCCCCCCOC(=O)CCCCCCCC(=O)OCC(F)(F)C(F)(F)C(F)(F)C(F)F

Mol. weight [g/mol]: 584.62

## Physical Properties

Property code	Value	Unit	Source
gf	-1843.78	kJ/mol	Joback Method
hf	-2690.62	kJ/mol	Joback Method
hfus	70.13	kJ/mol	Joback Method
hvap	83.20	kJ/mol	Joback Method
log10ws	-10.11		Crippen Method
logp	9.286		Crippen Method
mvol	420.330	ml/mol	McGowan Method
pc	620.96	kPa	Joback Method
rinpol	2793.00		NIST Webbook
rinpol	2793.00		NIST Webbook
tb	953.77	K	Joback Method
tc	1198.31	K	Joback Method
tf	535.35	K	Joback Method
vc	1.700	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1463.63	J/molxK	953.77	Joback Method
cpg	1485.97	J/molxK	994.53	Joback Method
cpg	1506.59	J/molxK	1035.28	Joback Method
cpg	1525.67	J/molxK	1076.04	Joback Method
cpg	1543.41	J/molxK	1116.80	Joback Method
cpg	1559.97	J/molxK	1157.56	Joback Method
cpg	1575.53	J/molxK	1198.31	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U355744&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U355744&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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