

# 3-Bromobenzoic acid, 3-methylphenyl ester

<b>Inchi:</b>	InChI=1S/C14H11BrO2/c1-10-4-2-7-13(8-10)17-14(16)11-5-3-6-12(15)9-11/h2-9H,1H3
<b>InchiKey:</b>	NMQQMNGWYHVNLO-UHFFFAOYSA-N
<b>Formula:</b>	C14H11BrO2
<b>SMILES:</b>	Cc1cccc(OC(=O)c2cccc(Br)c2)c1
<b>Mol. weight [g/mol]:</b>	291.14

## Physical Properties

Property code	Value	Unit	Source
gf	52.96	kJ/mol	Joback Method
hf	-100.64	kJ/mol	Joback Method
hfus	27.39	kJ/mol	Joback Method
hvap	68.22	kJ/mol	Joback Method
log10ws	-5.20		Crippen Method
logp	3.977		Crippen Method
mcvol	185.540	ml/mol	McGowan Method
pc	3035.62	kPa	Joback Method
rinpola	2008.00		NIST Webbook
rinpola	2008.00		NIST Webbook
tb	725.49	K	Joback Method
tc	980.66	K	Joback Method
tf	457.38	K	Joback Method
vc	0.690	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	448.93	J/molxK	725.49	Joback Method
cpg	503.71	J/molxK	938.13	Joback Method
cpg	494.80	J/molxK	895.60	Joback Method
cpg	484.93	J/molxK	853.07	Joback Method
cpg	474.03	J/molxK	810.55	Joback Method
cpg	462.05	J/molxK	768.02	Joback Method
cpg	511.71	J/molxK	980.66	Joback Method
dvisc	0.0001373	Paxs	725.49	Joback Method

dvisc	0.0001683	Paxs	680.81	Joback Method
dvisc	0.0002121	Paxs	636.12	Joback Method
dvisc	0.0002770	Paxs	591.43	Joback Method
dvisc	0.0003778	Paxs	546.75	Joback Method
dvisc	0.0005446	Paxs	502.06	Joback Method
dvisc	0.0008431	Paxs	457.38	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U307548&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U307548&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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