

1-Hydroxy-2-naphthamide

Inchi:	InChI=1S/C11H9NO2/c12-11(14)9-6-5-7-3-1-2-4-8(7)10(9)13/h1-6,13H,(H2,12,14)
InchiKey:	ZTXWIKHKNGFJAX-UHFFFAOYSA-N
Formula:	C11H9NO2
SMILES:	NC(=O)c1ccc2ccccc2c1O
Mol. weight [g/mol]:	187.19
CAS:	62353-80-4

Physical Properties

Property code	Value	Unit	Source
gf	34.08	kJ/mol	Joback Method
hf	-110.34	kJ/mol	Joback Method
hfus	27.50	kJ/mol	Joback Method
hvap	75.06	kJ/mol	Joback Method
log10ws	-2.99		Crippen Method
logp	1.644		Crippen Method
mcvol	140.050	ml/mol	McGowan Method
pc	4795.85	kPa	Joback Method
tb	708.74	K	Joback Method
tc	967.14	K	Joback Method
tf	530.28	K	Joback Method
vc	0.467	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	362.28	J/molxK	708.74	Joback Method
cpg	372.47	J/molxK	751.81	Joback Method
cpg	381.93	J/molxK	794.87	Joback Method
cpg	390.81	J/molxK	837.94	Joback Method
cpg	399.28	J/molxK	881.01	Joback Method
cpg	407.51	J/molxK	924.07	Joback Method
cpg	415.64	J/molxK	967.14	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C62353804&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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