

# 2,2',4,4',6,6'-Hexanitrobiphenyl

<b>Inchi:</b>	InChI=1S/C12H4N6O12/c19-13(20)5-1-7(15(23)24)11(8(2-5)16(25)26)12-9(17(27)28)3-6
<b>InchiKey:</b>	PJTZYMZXEQWUHG-UHFFFAOYSA-N
<b>Formula:</b>	C12H4N6O12
<b>SMILES:</b>	O=[N+]([O-])c1cc([N+](=O)[O-])c(-c2c([N+](=O)[O-])cc([N+](=O)[O-])cc2[N+](=O)[O-])c([N
<b>Mol. weight [g/mol]:</b>	424.19
<b>CAS:</b>	4433-16-3

## Physical Properties

Property code	Value	Unit	Source
chs	-5362.00 ± 4.00	kJ/mol	NIST Webbook
gf	430.50	kJ/mol	Joback Method
hf	48.67	kJ/mol	Joback Method
hfs	68.20 ± 8.40	kJ/mol	NIST Webbook
hfus	80.75	kJ/mol	Joback Method
hvap	150.38	kJ/mol	Joback Method
log10ws	-7.98		Crippen Method
logp	2.803		Crippen Method
mcvol	236.940	ml/mol	McGowan Method
pc	3356.75	kPa	Joback Method
tb	1468.24	K	Joback Method
tc	1809.37	K	Joback Method
tf	1214.62	K	Joback Method
vc	0.984	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	685.43	J/mol×K	1468.24	Joback Method
cpg	683.90	J/mol×K	1525.10	Joback Method
cpg	681.96	J/mol×K	1581.95	Joback Method
cpg	679.77	J/mol×K	1638.81	Joback Method
cpg	677.50	J/mol×K	1695.66	Joback Method
cpg	675.32	J/mol×K	1752.52	Joback Method
cpg	673.39	J/mol×K	1809.37	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4433163&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4433163&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

# Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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