

# Phthalic acid, decyl 2,2,3,3,4,4,4-heptafluorobutyl ester

<b>Inchi:</b>	InChI=1S/C22H27F7O4/c1-2-3-4-5-6-7-8-11-14-32-18(30)16-12-9-10-13-17(16)19(31)33
<b>InchiKey:</b>	YVIITHYTNXIEQY-UHFFFAOYSA-N
<b>Formula:</b>	C22H27F7O4
<b>SMILES:</b>	CCCCCCCCCOC(=O)c1ccccc1C(=O)OCC(F)(F)C(F)(F)C(F)(F)F
<b>Mol. weight [g/mol]:</b>	488.44

## Physical Properties

Property code	Value	Unit	Source
gf	-1585.85	kJ/mol	Joback Method
hf	-2160.97	kJ/mol	Joback Method
hfus	51.28	kJ/mol	Joback Method
hvap	76.21	kJ/mol	Joback Method
log10ws	-8.27		Crippen Method
logp	6.974		Crippen Method
mvol	324.350	ml/mol	McGowan Method
pc	988.88	kPa	Joback Method
rinpol	2317.00		NIST Webbook
rinpol	2317.00		NIST Webbook
tb	872.20	K	Joback Method
tc	1067.96	K	Joback Method
tf	532.35	K	Joback Method
vc	1.300	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1049.79	J/molxK	872.20	Joback Method
cpg	1064.46	J/molxK	904.83	Joback Method
cpg	1078.11	J/molxK	937.45	Joback Method
cpg	1090.81	J/molxK	970.08	Joback Method
cpg	1102.65	J/molxK	1002.71	Joback Method
cpg	1113.70	J/molxK	1035.34	Joback Method
cpg	1124.05	J/molxK	1067.96	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U415548&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U415548&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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