

# Succinic acid, 4-chloro-3-methylphenyl tetrahydrofurfuryl ester

Inchi:	InChI=1S/C16H19ClO5/c1-11-9-12(4-5-14(11)17)22-16(19)7-6-15(18)21-10-13-3-2-8-20
InchiKey:	KNGJOBUBHAKVSN-UHFFFAOYSA-N
Formula:	C16H19ClO5
SMILES:	<chem>Cc1cc(OC(=O)CCC(=O)OCC2CCCO2)ccc1Cl</chem>
Mol. weight [g/mol]:	326.77

## Physical Properties

Property code	Value	Unit	Source
gf	-352.35	kJ/mol	Joback Method
hf	-736.84	kJ/mol	Joback Method
hfus	42.14	kJ/mol	Joback Method
hvap	82.27	kJ/mol	Joback Method
log10ws	-3.83		Crippen Method
logp	3.056		Crippen Method
mvol	234.670	ml/mol	McGowan Method
pc	2016.32	kPa	Joback Method
rinpol	2593.00		NIST Webbook
rinpol	2593.00		NIST Webbook
tb	834.36	K	Joback Method
tc	1059.92	K	Joback Method
tf	533.25	K	Joback Method
vc	0.882	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	694.20	J/molxK	834.36	Joback Method
cpg	708.13	J/molxK	871.95	Joback Method
cpg	720.80	J/molxK	909.55	Joback Method
cpg	732.25	J/molxK	947.14	Joback Method
cpg	742.49	J/molxK	984.74	Joback Method
cpg	751.54	J/molxK	1022.33	Joback Method
cpg	759.42	J/molxK	1059.92	Joback Method
dvisc	0.0007159	Paxs	533.25	Joback Method

dvisc	0.0004562	Paxs	583.43	Joback Method
dvisc	0.0003122	Paxs	633.62	Joback Method
dvisc	0.0002259	Paxs	683.80	Joback Method
dvisc	0.0001708	Paxs	733.99	Joback Method
dvisc	0.0001339	Paxs	784.17	Joback Method
dvisc	0.0001081	Paxs	834.36	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U390724&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U390724&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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