

# Nonanoic acid, 2,2,2-trichloroethyl ester

<b>Other names:</b>	2,2,2-Trichloroethyl nonanoate
<b>Inchi:</b>	InChI=1S/C11H19Cl3O2/c1-2-3-4-5-6-7-8-10(15)16-9-11(12,13)14/h2-9H2,1H3
<b>InchiKey:</b>	UNSPRHCUPSMXLE-UHFFFAOYSA-N
<b>Formula:</b>	C11H19Cl3O2
<b>SMILES:</b>	CCCCCCCCC(=O)OCC(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	289.63

## Physical Properties

Property code	Value	Unit	Source
gf	-225.13	kJ/mol	Joback Method
hf	-571.14	kJ/mol	Joback Method
hfus	32.21	kJ/mol	Joback Method
hvap	61.09	kJ/mol	Joback Method
log10ws	-4.85		Crippen Method
logp	4.650		Crippen Method
mcvol	210.010	ml/mol	McGowan Method
pc	1857.91	kPa	Joback Method
rinpol	1694.00		NIST Webbook
rinpol	1660.00		NIST Webbook
rinpol	1675.00		NIST Webbook
rinpol	1674.00		NIST Webbook
rinpol	1678.00		NIST Webbook
rinpol	1660.00		NIST Webbook
rinpol	1676.00		NIST Webbook
rinpol	1679.00		NIST Webbook
rinpol	1679.00		NIST Webbook
rinpol	1678.00		NIST Webbook
rinpol	1694.00		NIST Webbook
ripol	2058.00		NIST Webbook
ripol	2045.00		NIST Webbook
ripol	2040.00		NIST Webbook
ripol	2090.00		NIST Webbook
ripol	2026.00		NIST Webbook
ripol	2039.00		NIST Webbook
ripol	2039.00		NIST Webbook
ripol	2026.00		NIST Webbook
ripol	2090.00		NIST Webbook

ripol	2065.00		NIST Webbook
tb	636.43	K	Joback Method
tc	830.26	K	Joback Method
tf	378.07	K	Joback Method
vc	0.811	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	507.77	J/molxK	636.43	Joback Method
cpg	567.74	J/molxK	797.96	Joback Method
cpg	557.15	J/molxK	765.65	Joback Method
cpg	545.89	J/molxK	733.35	Joback Method
cpg	533.93	J/molxK	701.04	Joback Method
cpg	521.23	J/molxK	668.74	Joback Method
cpg	577.68	J/molxK	830.26	Joback Method
dvisc	0.0001454	Paxs	636.43	Joback Method
dvisc	0.0001927	Paxs	593.37	Joback Method
dvisc	0.0002668	Paxs	550.31	Joback Method
dvisc	0.0003905	Paxs	507.25	Joback Method
dvisc	0.0006134	Paxs	464.19	Joback Method
dvisc	0.0010567	Paxs	421.13	Joback Method
dvisc	0.0020603	Paxs	378.07	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U360667&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U360667&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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