

# 1,2-Cyclohexanedicarboxylic acid, hexyl 2,4,6-trichlorophenyl ester

<b>Inchi:</b>	InChI=1S/C20H25Cl3O4/c1-2-3-4-7-10-26-19(24)14-8-5-6-9-15(14)20(25)27-18-16(22)1
<b>InchiKey:</b>	MUVXKQDJIMUPLM-UHFFFAOYSA-N
<b>Formula:</b>	C20H25Cl3O4
<b>SMILES:</b>	CCCCCOC(=O)C1CCCCC1C(=O)Oc1c(Cl)cc(Cl)cc1Cl
<b>Mol. weight [g/mol]:</b>	435.77

## Physical Properties

Property code	Value	Unit	Source
gf	-285.85	kJ/mol	Joback Method
hf	-756.85	kJ/mol	Joback Method
hfus	51.50	kJ/mol	Joback Method
hvap	95.96	kJ/mol	Joback Method
log10ws	-7.14		Crippen Method
logp	6.482		Crippen Method
mvol	309.640	ml/mol	McGowan Method
pc	1372.76	kPa	Joback Method
rinpol	2868.00		NIST Webbook
rinpol	2868.00		NIST Webbook
tb	978.37	K	Joback Method
tc	1211.12	K	Joback Method
tf	616.36	K	Joback Method
vc	1.175	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	948.09	J/molxK	978.37	Joback Method
cpg	992.85	J/molxK	1172.33	Joback Method
cpg	987.09	J/molxK	1133.54	Joback Method
cpg	979.75	J/molxK	1094.75	Joback Method
cpg	970.82	J/molxK	1055.95	Joback Method
cpg	960.27	J/molxK	1017.16	Joback Method
cpg	997.06	J/molxK	1211.12	Joback Method
dvisc	0.0000488	Paxs	978.37	Joback Method

dvisc	0.0000605	Paxs	918.04	Joback Method
dvisc	0.0000773	Paxs	857.70	Joback Method
dvisc	0.0001025	Paxs	797.37	Joback Method
dvisc	0.0001424	Paxs	737.03	Joback Method
dvisc	0.0002098	Paxs	676.70	Joback Method
dvisc	0.0003333	Paxs	616.36	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339811&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339811&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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