

# Fumaric acid, 3-chlorophenyl 2,2,3,3,4,4,5,5-octafluoropentyl ester

**Inchi:** InChI=1S/C15H9ClF8O4/c16-8-2-1-3-9(6-8)28-11(26)5-4-10(25)27-7-13(19,20)15(23,24)

**InchiKey:** IHFHMNZRHFIVBP-SNAWJCMRSA-N

**Formula:** C15H9ClF8O4

**SMILES:** O=C(C=CC(=O)Oc1cccc(Cl)c1)OCC(F)(F)C(F)(F)C(F)(F)C(F)F

**Mol. weight [g/mol]:** 440.67

## Physical Properties

Property code	Value	Unit	Source
gf	-1773.75	kJ/mol	Joback Method
hf	-2116.40	kJ/mol	Joback Method
hfus	37.11	kJ/mol	Joback Method
hvap	63.77	kJ/mol	Joback Method
log10ws	-5.37		Crippen Method
logp	4.516		Crippen Method
mvol	235.430	ml/mol	McGowan Method
pc	1562.28	kPa	Joback Method
rinpol	1916.00		NIST Webbook
rinpol	1916.00		NIST Webbook
tb	752.46	K	Joback Method
tc	940.99	K	Joback Method
tf	463.89	K	Joback Method
vc	0.950	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	664.69	J/molxK	752.46	Joback Method
cpg	674.81	J/molxK	783.88	Joback Method
cpg	684.11	J/molxK	815.30	Joback Method
cpg	692.67	J/molxK	846.73	Joback Method
cpg	700.54	J/molxK	878.15	Joback Method
cpg	707.80	J/molxK	909.57	Joback Method
cpg	714.51	J/molxK	940.99	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405851&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405851&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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