

# L-Valine, N-heptafluorobutyryl-, n-butyl ester

<b>Other names:</b>	L-Valine, N-heptafluorobutyryl-, isobutyl ester
<b>Inchi:</b>	InChI=1S/C13H18F7NO3/c1-4-5-6-24-9(22)8(7(2)3)21-10(23)11(14,15)12(16,17)13(18,19)20
<b>InchiKey:</b>	AOFDHKDMSRIGGQ-UHFFFAOYSA-N
<b>Formula:</b>	C13H18F7NO3
<b>SMILES:</b>	CCCCOC(=O)C(NC(=O)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(C)C
<b>Mol. weight [g/mol]:</b>	369.28

## Physical Properties

Property code	Value	Unit	Source
gf	-1574.90	kJ/mol	Joback Method
hf	-2025.14	kJ/mol	Joback Method
hfus	31.18	kJ/mol	Joback Method
hvap	56.49	kJ/mol	Joback Method
log10ws	-4.25		Crippen Method
logp	3.303		Crippen Method
mcvol	225.410	ml/mol	McGowan Method
pc	1495.35	kPa	Joback Method
rinpol	1344.00		NIST Webbook
rinpol	1307.00		NIST Webbook
rinpol	1307.00		NIST Webbook
tb	661.49	K	Joback Method
tc	827.12	K	Joback Method
tf	392.41	K	Joback Method
vc	0.909	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	662.12	J/molxK	661.49	Joback Method
cpg	675.32	J/molxK	689.10	Joback Method
cpg	687.71	J/molxK	716.70	Joback Method
cpg	699.34	J/molxK	744.31	Joback Method
cpg	710.26	J/molxK	771.91	Joback Method
cpg	720.49	J/molxK	799.52	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U320896&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U320896&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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