

# (E)-1-(4-Methoxy-3-nitro-phenyl)ethanone methoxime

<b>Other names:</b>	E-1-(4-Methoxy-3-nitrophenyl)-N-methoxy-ethanimine
<b>Inchi:</b>	InChI=1S/C10H12N2O4/c1-7(11-16-3)8-4-5-10(15-2)9(6-8)12(13)14/h4-6H,1-3H3
<b>InchiKey:</b>	KDBIODILSHUMOL-UHFFFAOYSA-N
<b>Formula:</b>	C10H12N2O4
<b>SMILES:</b>	CON=C(C)c1ccc(OC)c([N+](=O)[O-])c1
<b>Mol. weight [g/mol]:</b>	224.21

## Physical Properties

Property code	Value	Unit	Source
hf	-238.91	kJ/mol	Joback Method
hvap	66.26	kJ/mol	Joback Method
log10ws	-2.81		Crippen Method
logp	1.974		Crippen Method
mcvol	162.840	ml/mol	McGowan Method
pc	2502.50	kPa	Joback Method
rinpol	1878.00		NIST Webbook
rinpol	1878.00		NIST Webbook
tb	738.08	K	Joback Method
tc	983.92	K	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U373328&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U373328&amp;Units=SI</a>

## Legend

hf: Enthalpy of formation at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature

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