

# Dibenzo-p-dioxin, 1,2,3,4,7-pentachloro-

<b>Other names:</b>	1,2,3,4,7-pentachloro dibenzo-p-dioxin
<b>Inchi:</b>	InChI=1S/C12H3Cl5O2/c13-4-1-2-5-6(3-4)19-12-10(17)8(15)7(14)9(16)11(12)18-5/h1-3H
<b>InchiKey:</b>	WRNGAZFESPEMCN-UHFFFAOYSA-N
<b>Formula:</b>	C12H3Cl5O2
<b>SMILES:</b>	Clc1ccc2c(c1)Oc1c(Cl)c(Cl)c(Cl)c(Cl)c1O2
<b>Mol. weight [g/mol]:</b>	356.42
<b>CAS:</b>	39227-61-7

## Physical Properties

Property code	Value	Unit	Source
gf	56.24	kJ/mol	Joback Method
hf	-141.64	kJ/mol	Joback Method
hfus	48.30	kJ/mol	Joback Method
hvap	82.49	kJ/mol	Joback Method
log10ws	-6.41		Crippen Method
logp	6.852		Crippen Method
mcvol	194.500	ml/mol	McGowan Method
pc	2835.36	kPa	Joback Method
rinpol	2573.00		NIST Webbook
rinpol	2583.00		NIST Webbook
rinpol	2582.00		NIST Webbook
rinpol	2573.00		NIST Webbook
rinpol	2573.00		NIST Webbook
rinpol	2576.00		NIST Webbook
rinpol	2573.00		NIST Webbook
rinpol	2576.00		NIST Webbook
tb	810.37	K	Joback Method
tc	1079.48	K	Joback Method
tf	593.92	K	Joback Method
vc	0.745	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	404.85	J/molxK	810.37	Joback Method
cpg	411.89	J/molxK	855.22	Joback Method
cpg	418.44	J/molxK	900.07	Joback Method
cpg	424.59	J/molxK	944.92	Joback Method
cpg	430.47	J/molxK	989.77	Joback Method
cpg	436.17	J/molxK	1034.63	Joback Method
cpg	441.79	J/molxK	1079.48	Joback Method
dvisc	0.0010376	Paxs	593.92	Joback Method
dvisc	0.0008575	Paxs	630.00	Joback Method
dvisc	0.0007235	Paxs	666.07	Joback Method
dvisc	0.0006211	Paxs	702.14	Joback Method
dvisc	0.0005413	Paxs	738.22	Joback Method
dvisc	0.0004778	Paxs	774.30	Joback Method
dvisc	0.0004264	Paxs	810.37	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C39227617&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C39227617&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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