

# Benzoic acid, 3-amino-4-propoxy-, 2-(diethylamino)ethyl ester

Other names:

3-Amino-4-propoxybenzoic acid 2-(diethylamino)ethyl ester

2-(Diethylamino)ethyl 3-amino-4-propoxybenzoate

«beta»-(Diethylamino)ethyl 4-n-propoxybenzoate

Proparacaine

Proxymetacaine

Inchi: InChI=1S/C16H26N2O3/c1-4-10-20-15-8-7-13(12-14(15)17)16(19)21-11-9-18(5-2)6-3/h7

InchiKey: KCLANYCVBBTKTO-UHFFFAOYSA-N

Formula: C16H26N2O3

SMILES: CCCOc1ccc(C(=O)OCCN(CC)CC)cc1N

Mol. weight [g/mol]: 294.39

CAS: 499-67-2

## Physical Properties

Property code	Value	Unit	Source
gf	15.30	kJ/mol	Joback Method
hf	-435.68	kJ/mol	Joback Method
hfus	42.65	kJ/mol	Joback Method
hvap	79.06	kJ/mol	Joback Method
log10ws	-2.97		Crippen Method
logp	2.556		Crippen Method
mcvol	245.810	ml/mol	McGowan Method
pc	1781.84	kPa	Joback Method
rinpol	2323.00		NIST Webbook
rinpol	2323.00		NIST Webbook
tb	785.80	K	Joback Method
tc	988.00	K	Joback Method
tf	531.66	K	Joback Method
vc	0.912	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	737.96	J/mol×K	785.80	Joback Method
cpg	753.56	J/mol×K	819.50	Joback Method

cpg	768.14	J/mol×K	853.20	Joback Method
cpg	781.72	J/mol×K	886.90	Joback Method
cpg	794.31	J/mol×K	920.60	Joback Method
cpg	805.94	J/mol×K	954.30	Joback Method
cpg	816.62	J/mol×K	988.00	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C499672&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C499672&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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