

# 1,2,4-Trichloro-5-(decyloxy)benzene

<b>Inchi:</b>	InChI=1S/C16H23Cl3O/c1-2-3-4-5-6-7-8-9-10-20-16-12-14(18)13(17)11-15(16)19/h11-12
<b>InchiKey:</b>	FBFDILWRMGPTJU-UHFFFAOYSA-N
<b>Formula:</b>	C16H23Cl3O
<b>SMILES:</b>	CCCCCCCCCOc1cc(Cl)c(Cl)cc1Cl
<b>Mol. weight [g/mol]:</b>	337.71
<b>CAS:</b>	116401-38-8

## Physical Properties

Property code	Value	Unit	Source
gf	26.57	kJ/mol	Joback Method
hf	-350.89	kJ/mol	Joback Method
hfus	43.85	kJ/mol	Joback Method
hvap	71.04	kJ/mol	Joback Method
log10ws	-7.41		Crippen Method
logp	7.166		Crippen Method
mcvol	255.130	ml/mol	McGowan Method
pc	1483.85	kPa	Joback Method
tb	741.81	K	Joback Method
tc	944.15	K	Joback Method
tf	446.05	K	Joback Method
vc	0.989	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	665.96	J/molxK	741.81	Joback Method
cpg	681.10	J/molxK	775.53	Joback Method
cpg	695.35	J/molxK	809.26	Joback Method
cpg	708.75	J/molxK	842.98	Joback Method
cpg	721.30	J/molxK	876.70	Joback Method
cpg	733.04	J/molxK	910.43	Joback Method
cpg	743.99	J/molxK	944.15	Joback Method
dvisc	0.0007144	Paxs	446.05	Joback Method
dvisc	0.0004245	Paxs	495.34	Joback Method

dvisc	0.0002772	Paxs	544.64	Joback Method
dvisc	0.0001943	Paxs	593.93	Joback Method
dvisc	0.0001438	Paxs	643.22	Joback Method
dvisc	0.0001111	Paxs	692.52	Joback Method
dvisc	0.0000888	Paxs	741.81	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C116401388&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C116401388&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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