

# Succinic acid, 3-chlorophenyl (2-chlorocyclohexyl)methyl ester

**Inchi:** InChI=1S/C17H20Cl2O4/c18-13-5-3-6-14(10-13)23-17(21)9-8-16(20)22-11-12-4-1-2-7-15  
**InchiKey:** GMIRTKNNDZWBBM-UHFFFAOYSA-N  
**Formula:** C17H20Cl2O4  
**SMILES:** O=C(CCC(=O)Oc1cccc(Cl)c1)OCC1CCCCC1Cl  
**Mol. weight [g/mol]:** 359.24

## Physical Properties

Property code	Value	Unit	Source
gf	-279.92	kJ/mol	Joback Method
hf	-656.25	kJ/mol	Joback Method
hfus	40.31	kJ/mol	Joback Method
hvap	83.58	kJ/mol	Joback Method
log10ws	-5.02		Crippen Method
logp	4.366		Crippen Method
mvol	255.130	ml/mol	McGowan Method
pc	1820.06	kPa	Joback Method
rinpol	2742.00		NIST Webbook
rinpol	2742.00		NIST Webbook
tb	862.34	K	Joback Method
tc	1094.19	K	Joback Method
tf	527.59	K	Joback Method
vc	0.958	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	753.98	J/molxK	862.34	Joback Method
cpg	768.27	J/molxK	900.98	Joback Method
cpg	781.07	J/molxK	939.62	Joback Method
cpg	792.39	J/molxK	978.26	Joback Method
cpg	802.26	J/molxK	1016.91	Joback Method
cpg	810.71	J/molxK	1055.55	Joback Method
cpg	817.74	J/molxK	1094.19	Joback Method
dvisc	0.0006832	Paxs	527.59	Joback Method

dvisc	0.0004076	Paxs	583.38	Joback Method
dvisc	0.0002661	Paxs	639.17	Joback Method
dvisc	0.0001861	Paxs	694.96	Joback Method
dvisc	0.0001372	Paxs	750.76	Joback Method
dvisc	0.0001055	Paxs	806.55	Joback Method
dvisc	0.0000840	Paxs	862.34	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391403&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391403&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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