

5-Bromovaleric acid, 1-adamantylmethyl ester

Inchi:	InChI=1S/C16H25BrO2/c17-4-2-1-3-15(18)19-11-16-8-12-5-13(9-16)7-14(6-12)10-16/h1
InchiKey:	LDMGZPZQMNUSFN-UHFFFAOYSA-N
Formula:	C16H25BrO2
SMILES:	O=C(CCCCBBr)OCC12CC3CC(CC(C3)C1)C2
Mol. weight [g/mol]:	329.27

Physical Properties

Property code	Value	Unit	Source
gf	21.19	kJ/mol	Joback Method
hf	-384.90	kJ/mol	Joback Method
hfus	32.35	kJ/mol	Joback Method
hvap	65.25	kJ/mol	Joback Method
log10ws	-4.53		Crippen Method
logp	4.311		Crippen Method
mvol	228.660	ml/mol	McGowan Method
pc	2043.76	kPa	Joback Method
rinpol	2212.00		NIST Webbook
rinpol	2212.00		NIST Webbook
tb	727.99	K	Joback Method
tc	946.92	K	Joback Method
tf	472.00	K	Joback Method
vc	0.877	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	684.97	J/molxK	727.99	Joback Method
cpg	704.03	J/molxK	764.48	Joback Method
cpg	722.28	J/molxK	800.97	Joback Method
cpg	739.92	J/molxK	837.45	Joback Method
cpg	757.15	J/molxK	873.94	Joback Method
cpg	774.18	J/molxK	910.43	Joback Method
cpg	791.20	J/molxK	946.92	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U299982&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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