

# L-Valine, N-(5-fluoro-2-trifluoromethyl)-, pentyl ester

Inchi:	InChI=1S/C18H23F4NO3/c1-4-5-6-9-26-17(25)15(11(2)3)23-16(24)13-10-12(19)7-8-14(1
InchiKey:	AFLJIOVWFKRJEJ-UHFFFAOYSA-N
Formula:	C18H23F4NO3
SMILES:	CCCCCOC(=O)C(NC(=O)c1cc(F)ccc1C(F)(F)F)C(C)C
Mol. weight [g/mol]:	377.37

## Physical Properties

Property code	Value	Unit	Source
gf	-860.90	kJ/mol	Joback Method
hf	-1308.92	kJ/mol	Joback Method
hfus	42.98	kJ/mol	Joback Method
hvap	76.26	kJ/mol	Joback Method
log10ws	-5.75		Crippen Method
logp	4.332		Crippen Method
mcvol	266.790	ml/mol	McGowan Method
pc	1404.84	kPa	Joback Method
rinpol	2055.00		NIST Webbook
rinpol	2055.00		NIST Webbook
tb	821.18	K	Joback Method
tc	1015.63	K	Joback Method
tf	493.61	K	Joback Method
vc	1.050	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	822.16	J/molxK	821.18	Joback Method
cpg	836.09	J/molxK	853.59	Joback Method
cpg	849.06	J/molxK	886.00	Joback Method
cpg	861.10	J/molxK	918.40	Joback Method
cpg	872.26	J/molxK	950.81	Joback Method
cpg	882.58	J/molxK	983.22	Joback Method
cpg	892.11	J/molxK	1015.63	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U346423&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U346423&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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