

# Succinic acid, 2,2-dichloroethyl 3-methoxyphenyl ester

<b>Inchi:</b>	InChI=1S/C13H14Cl2O5/c1-18-9-3-2-4-10(7-9)20-13(17)6-5-12(16)19-8-11(14)15/h2-4,7
<b>InchiKey:</b>	CRGPFQYURXKDME-UHFFFAOYSA-N
<b>Formula:</b>	C13H14Cl2O5
<b>SMILES:</b>	COc1cccc(OC(=O)CCC(=O)OCC(Cl)Cl)c1
<b>Mol. weight [g/mol]:</b>	321.15

## Physical Properties

Property code	Value	Unit	Source
gf	-437.78	kJ/mol	Joback Method
hf	-745.17	kJ/mol	Joback Method
hfus	34.71	kJ/mol	Joback Method
hvap	76.57	kJ/mol	Joback Method
log10ws	-3.36		Crippen Method
logp	2.728		Crippen Method
mvol	215.500	ml/mol	McGowan Method
pc	2181.56	kPa	Joback Method
rinpol	2294.00		NIST Webbook
rinpol	2294.00		NIST Webbook
tb	777.92	K	Joback Method
tc	994.75	K	Joback Method
tf	486.60	K	Joback Method
vc	0.814	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	562.75	J/molxK	777.92	Joback Method
cpg	574.44	J/molxK	814.06	Joback Method
cpg	585.15	J/molxK	850.20	Joback Method
cpg	594.89	J/molxK	886.34	Joback Method
cpg	603.64	J/molxK	922.47	Joback Method
cpg	611.40	J/molxK	958.61	Joback Method
cpg	618.16	J/molxK	994.75	Joback Method
dvisc	0.0006326	Paxs	486.60	Joback Method

dvisc	0.0003784	Paxs	535.15	Joback Method
dvisc	0.0002466	Paxs	583.71	Joback Method
dvisc	0.0001716	Paxs	632.26	Joback Method
dvisc	0.0001257	Paxs	680.81	Joback Method
dvisc	0.0000960	Paxs	729.37	Joback Method
dvisc	0.0000759	Paxs	777.92	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U390979&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U390979&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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