

# Ethanamine, N-ethyl-N-nitro-

<b>Other names:</b>	Diethylamine, N-nitro- Diethylnitramine N-Nitrodiethylamine
<b>Inchi:</b>	InChI=1S/C4H10N2O2/c1-3-5(4-2)6(7)8/h3-4H2,1-2H3
<b>InchiKey:</b>	QAXAHXNXXIWQIZ-UHFFFAOYSA-N
<b>Formula:</b>	C4H10N2O2
<b>SMILES:</b>	CCN(CC)[N+](=O)[O-]
<b>Mol. weight [g/mol]:</b>	118.13
<b>CAS:</b>	7119-92-8

## Physical Properties

Property code	Value	Unit	Source
chl	-2897.00 ± 8.40	kJ/mol	NIST Webbook
chl	-2896.00	kJ/mol	NIST Webbook
gf	129.13	kJ/mol	Joback Method
hf	-69.12	kJ/mol	Joback Method
hfl	-106.00 ± 8.40	kJ/mol	NIST Webbook
hfus	20.50	kJ/mol	Joback Method
hvap	43.13	kJ/mol	Joback Method
log10ws	-1.14		Crippen Method
logp	0.520		Crippen Method
mcvol	94.620	ml/mol	McGowan Method
pc	3945.61	kPa	Joback Method
tb	455.20	K	Joback Method
tc	659.57	K	Joback Method
tf	310.92	K	Joback Method
vc	0.359	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	199.50	J/mol×K	455.20	Joback Method
cpg	209.93	J/mol×K	489.26	Joback Method
cpg	219.79	J/mol×K	523.32	Joback Method

cpg	229.10	J/mol×K	557.39	Joback Method
cpg	237.88	J/mol×K	591.45	Joback Method
cpg	246.15	J/mol×K	625.51	Joback Method
cpg	253.93	J/mol×K	659.57	Joback Method
hvapt	49.70	kJ/mol	358.00	NIST Webbook
hvapt	53.10 ± 3.30	kJ/mol	338.00	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7119928&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7119928&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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