

Phthalic acid, isobutyl 2,4,6-trichlorobenzyl ester

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|-----------------------------|--|
| Inchi: | InChI=1S/C19H17Cl3O4/c1-11(2)9-25-18(23)13-5-3-4-6-14(13)19(24)26-10-15-16(21)7- |
| InchiKey: | SJRTYIOATZJLJK-UHFFFAOYSA-N |
| Formula: | C19H17Cl3O4 |
| SMILES: | CC(C)COC(=O)c1cccc1C(=O)OCc1c(Cl)cc(Cl)cc1Cl |
| Mol. weight [g/mol]: | 415.69 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|---------|----------------|
| gf | -210.67 | kJ/mol | Joback Method |
| hf | -550.41 | kJ/mol | Joback Method |
| hfus | 46.13 | kJ/mol | Joback Method |
| hvap | 96.17 | kJ/mol | Joback Method |
| log10ws | -7.14 | | Crippen Method |
| logp | 5.817 | | Crippen Method |
| mvol | 282.650 | ml/mol | McGowan Method |
| pc | 1664.61 | kPa | Joback Method |
| rinpol | 2708.00 | | NIST Webbook |
| rinpol | 2708.00 | | NIST Webbook |
| tb | 971.83 | K | Joback Method |
| tc | 1212.62 | K | Joback Method |
| tf | 625.89 | K | Joback Method |
| vc | 1.073 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 783.80 | J/molxK | 971.83 | Joback Method |
| cpg | 819.16 | J/molxK | 1172.49 | Joback Method |
| cpg | 814.64 | J/molxK | 1132.36 | Joback Method |
| cpg | 808.87 | J/molxK | 1092.22 | Joback Method |
| cpg | 801.82 | J/molxK | 1052.09 | Joback Method |
| cpg | 793.47 | J/molxK | 1011.96 | Joback Method |
| cpg | 822.46 | J/molxK | 1212.62 | Joback Method |
| dvisc | 0.0000398 | Paxs | 971.83 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0000492 | Paxs | 914.17 | Joback Method |
| dvisc | 0.0000625 | Paxs | 856.52 | Joback Method |
| dvisc | 0.0000821 | Paxs | 798.86 | Joback Method |
| dvisc | 0.0001127 | Paxs | 741.20 | Joback Method |
| dvisc | 0.0001631 | Paxs | 683.55 | Joback Method |
| dvisc | 0.0002528 | Paxs | 625.89 | Joback Method |

Sources

| | |
|------------------------|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U382877&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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