

# 5(4H)-Oxazolone, 2-phenyl-4-(phenylmethylene)-, (Z)-

Other names:

2-Oxazolin-5-one, 4-benzylidene-2-phenyl-  
4-Benzylidene-2-phenyl-5(4H)-oxazolone  
5(4H)-Oxazolone, 2-phenyl-4-(phenylmethylene)-  
4-Benzyliden-2-phenyl-4,5-dihydro-1,3-oxazol-5-on  
4-Benzylidene-2-phenyl-4,5-dihydro-1,3-oxazol-5-one  
4-Benzylidene-2-phenyl-1,3-oxazol-5(4H)-one  
2-Oxazolin-5-one, 4-benzylidene-2-phenyl-, (Z)-  
(Z)-4-Benzylidene-2-phenyloxazolin-5-one  
4-Benzylidene-2-phenyl-1,3-oxazol-5(4H)-one, cis-

**Inchi:** InChI=1S/C16H11NO2/c18-16-14(11-12-7-3-1-4-8-12)17-15(19-16)13-9-5-2-6-10-13/h1-

**InchiKey:** VFDOKJVMHZUBTN-SDNWHVSQSA-N

**Formula:** C16H11NO2

**SMILES:** O=C1OC(c2ccccc2)=NC1=Cc1ccccc1

**Mol. weight [g/mol]:** 249.26

**CAS:** 17606-70-1

## Physical Properties

Property code	Value	Unit	Source
chs	-7778.10	kJ/mol	NIST Webbook
gf	326.78	kJ/mol	Joback Method
hf	103.92	kJ/mol	Joback Method
hfus	31.93	kJ/mol	Joback Method
hvap	73.04	kJ/mol	Joback Method
log10ws	-3.76		Crippen Method
logp	3.031		Crippen Method
mcvol	186.740	ml/mol	McGowan Method
pc	3079.57	kPa	Joback Method
tb	798.04	K	Joback Method
tc	1083.82	K	Joback Method
tf	528.03	K	Joback Method
vc	0.704	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	536.50	J/mol×K	798.04	Joback Method
cpg	551.87	J/mol×K	845.67	Joback Method
cpg	565.40	J/mol×K	893.30	Joback Method
cpg	577.15	J/mol×K	940.93	Joback Method
cpg	587.18	J/mol×K	988.56	Joback Method
cpg	595.56	J/mol×K	1036.19	Joback Method
cpg	602.35	J/mol×K	1083.82	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C17606701&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C17606701&amp;Units=SI</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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