

# Isophthalic acid, 3-chlorophenyl ethyl ester

**Inchi:** InChI=1S/C16H13ClO4/c1-2-20-15(18)11-5-3-6-12(9-11)16(19)21-14-8-4-7-13(17)10-14/  
**InchiKey:** FLARZRNEIVBXL5-UHFFFAOYSA-N  
**Formula:** C16H13ClO4  
**SMILES:** CCOC(=O)c1cccc(C(=O)Oc2cccc(Cl)c2)c1  
**Mol. weight [g/mol]:** 304.73

## Physical Properties

Property code	Value	Unit	Source
gf	-190.37	kJ/mol	Joback Method
hf	-428.79	kJ/mol	Joback Method
hfus	34.27	kJ/mol	Joback Method
hvap	79.78	kJ/mol	Joback Method
log10ws	-4.90		Crippen Method
logp	3.736		Crippen Method
mvol	215.900	ml/mol	McGowan Method
pc	2340.56	kPa	Joback Method
rinpol	2446.00		NIST Webbook
rinpol	2446.00		NIST Webbook
tb	818.81	K	Joback Method
tc	1057.73	K	Joback Method
tf	522.20	K	Joback Method
vc	0.812	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	579.61	J/molxK	818.81	Joback Method
cpg	591.66	J/molxK	858.63	Joback Method
cpg	602.52	J/molxK	898.45	Joback Method
cpg	612.21	J/molxK	938.27	Joback Method
cpg	620.74	J/molxK	978.09	Joback Method
cpg	628.16	J/molxK	1017.91	Joback Method
cpg	634.47	J/molxK	1057.73	Joback Method
dvisc	0.0005504	Paxs	522.20	Joback Method

dvisc	0.0003526	Paxs	571.63	Joback Method
dvisc	0.0002425	Paxs	621.07	Joback Method
dvisc	0.0001762	Paxs	670.50	Joback Method
dvisc	0.0001338	Paxs	719.94	Joback Method
dvisc	0.0001052	Paxs	769.38	Joback Method
dvisc	0.0000852	Paxs	818.81	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344599&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344599&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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