

# Fumaric acid, 2-octyl 2,4,6-trichlorophenyl ester

Inchi:	InChI=1S/C18H21Cl3O4/c1-3-4-5-6-7-12(2)24-16(22)8-9-17(23)25-18-14(20)10-13(19)1
InchiKey:	YICOYUMTYGGFAX-CMDGGGOBGSA-N
Formula:	C18H21Cl3O4
SMILES:	CCCCCCC(C)OC(=O)C=CC(=O)Oc1c(Cl)cc(Cl)cc1Cl
Mol. weight [g/mol]:	407.72

## Physical Properties

Property code	Value	Unit	Source
gf	-241.65	kJ/mol	Joback Method
hf	-637.61	kJ/mol	Joback Method
hfus	50.09	kJ/mol	Joback Method
hvap	90.96	kJ/mol	Joback Method
log10ws	-6.85		Crippen Method
logp	6.010		Crippen Method
mcvol	288.020	ml/mol	McGowan Method
pc	1457.90	kPa	Joback Method
rinpol	2576.00		NIST Webbook
rinpol	2576.00		NIST Webbook
tb	921.45	K	Joback Method
tc	1142.93	K	Joback Method
tf	570.60	K	Joback Method
vc	1.105	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	806.41	J/molxK	921.45	Joback Method
cpg	818.15	J/molxK	958.36	Joback Method
cpg	828.83	J/molxK	995.28	Joback Method
cpg	838.49	J/molxK	1032.19	Joback Method
cpg	847.16	J/molxK	1069.10	Joback Method
cpg	854.86	J/molxK	1106.02	Joback Method
cpg	861.63	J/molxK	1142.93	Joback Method
dvisc	0.0003122	Paxs	570.60	Joback Method

dvisc	0.0001869	Paxs	629.08	Joback Method
dvisc	0.0001221	Paxs	687.55	Joback Method
dvisc	0.0000853	Paxs	746.02	Joback Method
dvisc	0.0000627	Paxs	804.50	Joback Method
dvisc	0.0000481	Paxs	862.97	Joback Method
dvisc	0.0000382	Paxs	921.45	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405593&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405593&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/116-081-3/Fumaric-acid-2-octyl-2-4-6-trichlorophenyl-ester.pdf>

Generated by Cheméo on 2024-04-30 18:33:07.616020027 +0000 UTC m=+16791236.536597339.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.