

Glutaric acid, 2,2,3,3,4,4,5,5-octafluoropentyl 3-fluorophenyl ester

Inchi:	InChI=1S/C16H13F9O4/c17-9-3-1-4-10(7-9)29-12(27)6-2-5-11(26)28-8-14(20,21)16(24,25)
InchiKey:	CWTWUJNSRFMKNC-UHFFFAOYSA-N
Formula:	C16H13F9O4
SMILES:	O=C(CCCC(=O)Oc1cccc(F)c1)OCC(F)(F)C(F)(F)C(F)(F)C(F)F
Mol. weight [g/mol]:	440.26

Physical Properties

Property code	Value	Unit	Source
gf	-2028.43	kJ/mol	Joback Method
hf	-2434.63	kJ/mol	Joback Method
hfus	38.38	kJ/mol	Joback Method
hvap	60.83	kJ/mol	Joback Method
log10ws	-5.58		Crippen Method
logp	4.616		Crippen Method
mcvol	243.350	ml/mol	McGowan Method
pc	1386.08	kPa	Joback Method
rinpol	1816.00		NIST Webbook
rinpol	1816.00		NIST Webbook
tb	733.02	K	Joback Method
tc	908.43	K	Joback Method
tf	450.91	K	Joback Method
vc	0.995	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	725.73	J/mol×K	733.02	Joback Method
cpg	737.57	J/mol×K	762.26	Joback Method
cpg	748.56	J/mol×K	791.49	Joback Method
cpg	758.76	J/mol×K	820.73	Joback Method
cpg	768.22	J/mol×K	849.96	Joback Method
cpg	776.98	J/mol×K	879.20	Joback Method
cpg	785.09	J/mol×K	908.43	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U392085&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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