

# Glutaric acid, 3-methylbut-2-en-1-yl 3-fluorophenyl ester

<b>Inchi:</b>	InChI=1S/C16H19FO4/c1-12(2)9-10-20-15(18)7-4-8-16(19)21-14-6-3-5-13(17)11-14/h3,5
<b>InchiKey:</b>	YGGJMAUDOGVZKJ-UHFFFAOYSA-N
<b>Formula:</b>	C16H19FO4
<b>SMILES:</b>	CC(C)=CCOC(=O)CCCC(=O)Oc1cccc(F)c1
<b>Mol. weight [g/mol]:</b>	294.32

## Physical Properties

Property code	Value	Unit	Source
gf	-404.36	kJ/mol	Joback Method
hf	-726.79	kJ/mol	Joback Method
hfus	38.39	kJ/mol	Joback Method
hvap	71.68	kJ/mol	Joback Method
log10ws	-4.18		Crippen Method
logp	3.411		Crippen Method
mcvol	224.890	ml/mol	McGowan Method
pc	1835.69	kPa	Joback Method
rinpola	2032.00		NIST Webbook
rinpola	2032.00		NIST Webbook
tb	753.03	K	Joback Method
tc	956.57	K	Joback Method
tf	434.89	K	Joback Method
vc	0.871	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	631.35	J/molxK	753.03	Joback Method
cpg	645.53	J/molxK	786.95	Joback Method
cpg	658.79	J/molxK	820.88	Joback Method
cpg	671.15	J/molxK	854.80	Joback Method
cpg	682.63	J/molxK	888.72	Joback Method
cpg	693.26	J/molxK	922.64	Joback Method
cpg	703.07	J/molxK	956.57	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U392089&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U392089&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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