

# Succinic acid, 4-chloro-3-methylphenyl hex-5-en-1-yl ester

<b>Inchi:</b>	InChI=1S/C17H21ClO4/c1-3-4-5-6-11-21-16(19)9-10-17(20)22-14-7-8-15(18)13(2)12-14
<b>InchiKey:</b>	UPDFPDFCLIGDGL-UHFFFAOYSA-N
<b>Formula:</b>	C17H21ClO4
<b>SMILES:</b>	C=CCCCOC(=O)CCC(=O)Oc1ccc(Cl)c(C)c1
<b>Mol. weight [g/mol]:</b>	324.80

## Physical Properties

Property code	Value	Unit	Source
gf	-206.52	kJ/mol	Joback Method
hf	-560.53	kJ/mol	Joback Method
hfus	41.54	kJ/mol	Joback Method
hvap	79.06	kJ/mol	Joback Method
log10ws	-5.01		Crippen Method
logp	4.234		Crippen Method
mvol	249.450	ml/mol	McGowan Method
pc	1663.26	kPa	Joback Method
rinpol	2420.00		NIST Webbook
rinpol	2420.00		NIST Webbook
tb	811.69	K	Joback Method
tc	1019.33	K	Joback Method
tf	505.29	K	Joback Method
vc	0.958	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	703.91	J/molxK	811.69	Joback Method
cpg	717.59	J/molxK	846.30	Joback Method
cpg	730.28	J/molxK	880.90	Joback Method
cpg	741.99	J/molxK	915.51	Joback Method
cpg	752.74	J/molxK	950.11	Joback Method
cpg	762.54	J/molxK	984.72	Joback Method
cpg	771.41	J/molxK	1019.33	Joback Method
dvisc	0.0005783	Paxs	505.29	Joback Method

dvisc	0.0003551	Paxs	556.36	Joback Method
dvisc	0.0002367	Paxs	607.42	Joback Method
dvisc	0.0001680	Paxs	658.49	Joback Method
dvisc	0.0001253	Paxs	709.56	Joback Method
dvisc	0.0000972	Paxs	760.62	Joback Method
dvisc	0.0000778	Paxs	811.69	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391289&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391289&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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