

# Propionamide, 2,3-dichloro-N-pentyl-

<b>Inchi:</b>	InChI=1S/C8H15Cl2NO/c1-2-3-4-5-11-8(12)7(10)6-9/h7H,2-6H2,1H3,(H,11,12)
<b>InchiKey:</b>	KZQJTJJPFSSQOBT-UHFFFAOYSA-N
<b>Formula:</b>	C8H15Cl2NO
<b>SMILES:</b>	CCCCCNC(=O)C(Cl)CCl
<b>Mol. weight [g/mol]:</b>	212.12

## Physical Properties

Property code	Value	Unit	Source
gf	-49.35	kJ/mol	Joback Method
hf	-304.32	kJ/mol	Joback Method
hfus	28.05	kJ/mol	Joback Method
hvap	54.97	kJ/mol	Joback Method
log10ws	-2.55		Crippen Method
logp	2.139		Crippen Method
mvol	159.610	ml/mol	McGowan Method
pc	2561.10	kPa	Joback Method
rinpol	1644.00		NIST Webbook
rinpol	1644.00		NIST Webbook
tb	560.90	K	Joback Method
tc	753.45	K	Joback Method
tf	327.35	K	Joback Method
vc	0.617	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	357.75	J/mol×K	560.90	Joback Method
cpg	369.77	J/mol×K	592.99	Joback Method
cpg	381.16	J/mol×K	625.08	Joback Method
cpg	391.95	J/mol×K	657.17	Joback Method
cpg	402.16	J/mol×K	689.26	Joback Method
cpg	411.81	J/mol×K	721.36	Joback Method
cpg	420.91	J/mol×K	753.45	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U415226&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U415226&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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