

# Succinic acid, 4-chloro-3-methylphenyl cyclopentyl ester

<b>Inchi:</b>	InChI=1S/C16H19ClO4/c1-11-10-13(6-7-14(11)17)21-16(19)9-8-15(18)20-12-4-2-3-5-12
<b>InchiKey:</b>	BKWSNQQNNAURIY-UHFFFAOYSA-N
<b>Formula:</b>	C16H19ClO4
<b>SMILES:</b>	<chem>Cc1cc(OC(=O)CCC(=O)OC2CCCC2)ccc1Cl</chem>
<b>Mol. weight [g/mol]:</b>	310.77

## Physical Properties

Property code	Value	Unit	Source
gf	-266.23	kJ/mol	Joback Method
hf	-604.84	kJ/mol	Joback Method
hfus	34.16	kJ/mol	Joback Method
hvap	77.76	kJ/mol	Joback Method
log10ws	-4.75		Crippen Method
logp	3.820		Crippen Method
mvol	228.800	ml/mol	McGowan Method
pc	2032.72	kPa	Joback Method
rinpol	2396.00		NIST Webbook
rinpol	2396.00		NIST Webbook
tb	807.41	K	Joback Method
tc	1034.61	K	Joback Method
tf	506.68	K	Joback Method
vc	0.862	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	664.05	J/molxK	807.41	Joback Method
cpg	678.91	J/molxK	845.28	Joback Method
cpg	692.51	J/molxK	883.14	Joback Method
cpg	704.87	J/molxK	921.01	Joback Method
cpg	716.02	J/molxK	958.88	Joback Method
cpg	725.98	J/molxK	996.74	Joback Method
cpg	734.78	J/molxK	1034.61	Joback Method
dvisc	0.0008159	Paxs	506.68	Joback Method

dvisc	0.0005171	Paxs	556.80	Joback Method
dvisc	0.0003534	Paxs	606.92	Joback Method
dvisc	0.0002560	Paxs	657.04	Joback Method
dvisc	0.0001940	Paxs	707.17	Joback Method
dvisc	0.0001526	Paxs	757.29	Joback Method
dvisc	0.0001237	Paxs	807.41	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391381&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391381&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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