2,4,5-Trifluoro-3-methoxybenzoic acid, 3-pentadecyl ester

Inchi: InChI=1S/C23H35F3O3/c1-4-6-7-8-9-10-11-12-13-14-15-17(5-2)29-23(27)18-16-19(24)2

InchiKey: OYBDQOJAPLBKNJ-UHFFFAOYSA-N

Formula: C23H35F3O3

SMILES: CCCCCCCCCCCCC(CC)OC(=O)c1cc(F)c(F)c(OC)c1F

Mol. weight [g/mol]: 416.52

Physical Properties

Property code	Value	Unit	Source
gf	-709.12	kJ/mol	Joback Method
hf	-1298.03	kJ/mol	Joback Method
hfus	57.50	kJ/mol	Joback Method
hvap	80.44	kJ/mol	Joback Method
log10ws	-8.80		Crippen Method
logp	7.359		Crippen Method
mcvol	329.790	ml/mol	McGowan Method
pc	945.00	kPa	Joback Method
rinpol	2409.00		NIST Webbook
rinpol	2409.00		NIST Webbook
tb	868.32	K	Joback Method
tc	1063.27	K	Joback Method
tf	506.63	K	Joback Method
vc	1.306	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1059.02	J/mol×K	868.32	Joback Method
cpg	1076.63	J/mol×K	900.81	Joback Method
cpg	1093.03	J/mol×K	933.30	Joback Method
cpg	1108.24	J/mol×K	965.79	Joback Method
cpg	1122.29	J/mol×K	998.28	Joback Method
cpg	1135.19	J/mol×K	1030.78	Joback Method
cpg	1146.96	J/mol×K	1063.27	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U338466&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume