

# Dibenzo-p-dioxin, 1,2,4,6,7,9-hexachloro-

<b>Other names:</b>	1,2,4,6,7,9-hexachloro dibenzo-p-dioxin
<b>Inchi:</b>	InChI=1S/C12H2Cl6O2/c13-3-1-5(15)9-11(7(3)17)20-10-6(16)2-4(14)8(18)12(10)19-9/h1
<b>InchiKey:</b>	BSJDQMWAFTDGD-UHFFFAOYSA-N
<b>Formula:</b>	C12H2Cl6O2
<b>SMILES:</b>	Clc1cc(Cl)c2c(c1Cl)Oc1c(Cl)cc(Cl)c(Cl)c1O2
<b>Mol. weight [g/mol]:</b>	390.86
<b>CAS:</b>	39227-62-8

## Physical Properties

Property code	Value	Unit	Source
gf	34.68	kJ/mol	Joback Method
hf	-168.85	kJ/mol	Joback Method
hfus	52.11	kJ/mol	Joback Method
hvap	87.53	kJ/mol	Joback Method
log10ws	-7.10		Crippen Method
logp	7.505		Crippen Method
mcvol	206.740	ml/mol	McGowan Method
pc	2684.64	kPa	Joback Method
rinpol	2732.00		NIST Webbook
rinpol	2731.00		NIST Webbook
rinpol	2713.00		NIST Webbook
rinpol	2713.00		NIST Webbook
rinpol	2668.00		NIST Webbook
rinpol	2713.00		NIST Webbook
rinpol	2725.00		NIST Webbook
rinpol	2668.00		NIST Webbook
rinpol	2713.00		NIST Webbook
tb	852.78	K	Joback Method
tc	1123.61	K	Joback Method
tf	636.36	K	Joback Method
vc	0.793	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	417.83	J/mol×K	852.78	Joback Method
cpg	424.09	J/mol×K	897.92	Joback Method
cpg	429.98	J/mol×K	943.06	Joback Method
cpg	435.58	J/mol×K	988.19	Joback Method
cpg	440.99	J/mol×K	1033.33	Joback Method
cpg	446.32	J/mol×K	1078.47	Joback Method
cpg	451.65	J/mol×K	1123.61	Joback Method
dvisc	0.0009011	Paxs	636.36	Joback Method
dvisc	0.0007551	Paxs	672.43	Joback Method
dvisc	0.0006442	Paxs	708.50	Joback Method
dvisc	0.0005581	Paxs	744.57	Joback Method
dvisc	0.0004900	Paxs	780.64	Joback Method
dvisc	0.0004352	Paxs	816.71	Joback Method
dvisc	0.0003904	Paxs	852.78	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C39227628&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C39227628&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307i">http://pubs.acs.org/doi/abs/10.1021/ci990307i</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/116-259-6/Dibenzo-p-dioxin-1-2-4-6-7-9-hexachloro.pdf>

Generated by Cheméo on 2024-04-17 22:59:19.634939605 +0000 UTC m=+15684008.555516915.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.