

Barbituric acid, 5-(2-propenyl)-5-(2-hydroxypropyl)

Inchi: InChI=1S/C10H14N2O4/c1-3-4-10(5-6(2)13)7(14)11-9(16)12-8(10)15/h3,6,13H,1,4-5H2,
InchiKey: VNLMRPAWAMPLNZ-UHFFFAOYSA-N
Formula: C10H14N2O4
SMILES: C=CCC1(CC(C)O)C(=O)NC(=O)NC1=O
Mol. weight [g/mol]: 226.23
CAS: 42013-22-9

Physical Properties

Property code	Value	Unit	Source
gf	-191.49	kJ/mol	Joback Method
hf	-549.73	kJ/mol	Joback Method
hfus	24.19	kJ/mol	Joback Method
hvap	79.01	kJ/mol	Joback Method
log10ws	-1.57		Crippen Method
logp	-0.314		Crippen Method
mcvol	167.140	ml/mol	McGowan Method
pc	3759.17	kPa	Joback Method
rinpol	1837.00		NIST Webbook
rinpol	1837.00		NIST Webbook
tb	836.97	K	Joback Method
tc	1077.73	K	Joback Method
tf	692.52	K	Joback Method
vc	0.616	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	524.14	J/molxK	836.97	Joback Method
cpg	538.60	J/molxK	877.10	Joback Method
cpg	552.29	J/molxK	917.22	Joback Method
cpg	565.23	J/molxK	957.35	Joback Method
cpg	577.43	J/molxK	997.48	Joback Method
cpg	588.90	J/molxK	1037.60	Joback Method
cpg	599.66	J/molxK	1077.73	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C42013229&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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