

# 5-Ethyl-5-(3-oxy-1-cyclohexenyl)-hexahydropyrim

**Other names:**

3'-Oxocyclobarbital  
Cyclobarbital, (3-oxo)-  
Barbituric acid, 5-ethyl-5-(3-oxo-1-cyclohexen-1-yl)-  
Cyclobarbital M (oxo)

**Inchi:** InChI=1S/C12H14N2O4/c1-2-12(7-4-3-5-8(15)6-7)9(16)13-11(18)14-10(12)17/h6H,2-5H2**InchiKey:** QECKKEAIYXGFLF-UHFFFAOYSA-N**Formula:** C12H14N2O4**SMILES:** CCC1(C2=CC(=O)CCC2)C(=O)NC(=O)NC1=O**Mol. weight [g/mol]:** 250.25**CAS:** 35305-10-3

## Physical Properties

Property code	Value	Unit	Source
gf	-193.33	kJ/mol	Joback Method
hf	-575.66	kJ/mol	Joback Method
hfus	21.19	kJ/mol	Joback Method
hvap	73.78	kJ/mol	Joback Method
log10ws	-2.21		Crippen Method
logp	0.428		Crippen Method
mcvol	180.160	ml/mol	McGowan Method
pc	3581.35	kPa	Joback Method
tb	890.49	K	Joback Method
tc	1183.07	K	Joback Method
tf	764.12	K	Joback Method
vc	0.660	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	596.20	J/molxK	890.49	Joback Method
cpg	614.08	J/molxK	939.25	Joback Method
cpg	630.18	J/molxK	988.02	Joback Method
cpg	644.45	J/molxK	1036.78	Joback Method
cpg	656.86	J/molxK	1085.54	Joback Method

cpg	667.35	J/mol×K	1134.30	Joback Method
cpg	675.88	J/mol×K	1183.07	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C35305103&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C35305103&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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