

Terephthalic acid, ethyl 2-iodobenzyl ester

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|-----------------------------|-----------------------------------------------------------------------------------|
| Inchi: | InChI=1S/C17H15IO4/c1-2-21-16(19)12-7-9-13(10-8-12)17(20)22-11-14-5-3-4-6-15(14)1 |
| InchiKey: | PVJLQXGRZFQJMT-UHFFFAOYSA-N |
| Formula: | C17H15IO4 |
| SMILES: | CCOC(=O)c1ccc(C(=O)OCc2ccccc2I)cc1 |
| Mol. weight [g/mol]: | 410.20 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -111.90 | kJ/mol | Joback Method |
| hf | -356.82 | kJ/mol | Joback Method |
| hfus | 37.07 | kJ/mol | Joback Method |
| hvap | 87.00 | kJ/mol | Joback Method |
| log10ws | -5.63 | | Crippen Method |
| logp | 3.825 | | Crippen Method |
| mvol | 243.570 | ml/mol | McGowan Method |
| pc | 2167.36 | kPa | Joback Method |
| rinpol | 2856.00 | | NIST Webbook |
| rinpol | 2856.00 | | NIST Webbook |
| tb | 897.40 | K | Joback Method |
| tc | 1150.26 | K | Joback Method |
| tf | 561.61 | K | Joback Method |
| vc | 0.907 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 652.28 | J/molxK | 897.40 | Joback Method |
| cpg | 696.29 | J/molxK | 1108.12 | Joback Method |
| cpg | 689.84 | J/molxK | 1065.97 | Joback Method |
| cpg | 682.25 | J/molxK | 1023.83 | Joback Method |
| cpg | 673.50 | J/molxK | 981.69 | Joback Method |
| cpg | 663.53 | J/molxK | 939.54 | Joback Method |
| cpg | 701.67 | J/molxK | 1150.26 | Joback Method |
| dvisc | 0.0000638 | Paxs | 897.40 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0000795 | Paxs | 841.43 | Joback Method |
| dvisc | 0.0001022 | Paxs | 785.47 | Joback Method |
| dvisc | 0.0001366 | Paxs | 729.50 | Joback Method |
| dvisc | 0.0001915 | Paxs | 673.54 | Joback Method |
| dvisc | 0.0002855 | Paxs | 617.58 | Joback Method |
| dvisc | 0.0004608 | Paxs | 561.61 | Joback Method |

Sources

| | |
|------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U416071&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.cheméo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |

Legend

| | |
|-----------------|-------------------------------------------------|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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