

# Butyric acid, 2-phenyl-, 3-chlorophenyl ester

<b>Inchi:</b>	InChI=1S/C16H15ClO2/c1-2-15(12-7-4-3-5-8-12)16(18)19-14-10-6-9-13(17)11-14/h3-11,
<b>InchiKey:</b>	OHIKIFPKDVGHPY-UHFFFAOYSA-N
<b>Formula:</b>	C16H15ClO2
<b>SMILES:</b>	CCC(C(=O)Oc1cccc(Cl)c1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	274.74

## Physical Properties

Property code	Value	Unit	Source
gf	50.74	kJ/mol	Joback Method
hf	-177.80	kJ/mol	Joback Method
hfus	28.35	kJ/mol	Joback Method
hvap	69.58	kJ/mol	Joback Method
log10ws	-4.89		Crippen Method
logp	4.439		Crippen Method
mvol	208.460	ml/mol	McGowan Method
pc	2289.32	kPa	Joback Method
rinpol	1998.00		NIST Webbook
rinpol	1998.00		NIST Webbook
tb	737.10	K	Joback Method
tc	977.90	K	Joback Method
tf	422.52	K	Joback Method
vc	0.782	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	544.33	J/molxK	737.10	Joback Method
cpg	607.05	J/molxK	937.77	Joback Method
cpg	596.76	J/molxK	897.64	Joback Method
cpg	585.40	J/molxK	857.50	Joback Method
cpg	572.91	J/molxK	817.37	Joback Method
cpg	559.24	J/molxK	777.23	Joback Method
cpg	616.32	J/molxK	977.90	Joback Method
dvisc	0.0001031	Paxs	737.10	Joback Method

dvisc	0.0001325	Paxs	684.67	Joback Method
dvisc	0.0001774	Paxs	632.24	Joback Method
dvisc	0.0002504	Paxs	579.81	Joback Method
dvisc	0.0003785	Paxs	527.38	Joback Method
dvisc	0.0006268	Paxs	474.95	Joback Method
dvisc	0.0011764	Paxs	422.52	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406863&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406863&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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