

# Succinic acid, 1-(3-bromophenyl)ethyl pentyl ester

Inchi:	InChI=1S/C17H23BrO4/c1-3-4-5-11-21-16(19)9-10-17(20)22-13(2)14-7-6-8-15(18)12-14
InchiKey:	INJAKOUIUQOKFOE-UHFFFAOYSA-N
Formula:	C17H23BrO4
SMILES:	CCCCCOC(=O)CCC(=O)OC(C)c1cccc(Br)c1
Mol. weight [g/mol]:	371.27

## Physical Properties

Property code	Value	Unit	Source
gf	-260.92	kJ/mol	Joback Method
hf	-637.70	kJ/mol	Joback Method
hfus	40.77	kJ/mol	Joback Method
hvap	80.73	kJ/mol	Joback Method
log10ws	-5.39		Crippen Method
logp	4.567		Crippen Method
mcvol	259.010	ml/mol	McGowan Method
pc	1787.88	kPa	Joback Method
rinpola	2326.00		NIST Webbook
rinpola	2326.00		NIST Webbook
tb	838.32	K	Joback Method
tc	1051.65	K	Joback Method
tf	509.41	K	Joback Method
vc	0.984	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	744.57	J/molxK	838.32	Joback Method
cpg	803.77	J/molxK	1016.09	Joback Method
cpg	793.97	J/molxK	980.54	Joback Method
cpg	783.18	J/molxK	944.98	Joback Method
cpg	771.36	J/molxK	909.43	Joback Method
cpg	758.50	J/molxK	873.87	Joback Method
cpg	812.60	J/molxK	1051.65	Joback Method
dvisc	0.0000612	Paxs	838.32	Joback Method

dvisc	0.0000782	Paxs	783.50	Joback Method
dvisc	0.0001039	Paxs	728.68	Joback Method
dvisc	0.0001444	Paxs	673.87	Joback Method
dvisc	0.0002129	Paxs	619.05	Joback Method
dvisc	0.0003384	Paxs	564.23	Joback Method
dvisc	0.0005943	Paxs	509.41	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U381465&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U381465&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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