

# Isophthalic acid, monochloride, hexyl ester

<b>Inchi:</b>	InChI=1S/C14H17ClO3/c1-2-3-4-5-9-18-14(17)12-8-6-7-11(10-12)13(15)16/h6-8,10H,2-5
<b>InchiKey:</b>	BIOOGHNAZHAIJW-UHFFFAOYSA-N
<b>Formula:</b>	C14H17ClO3
<b>SMILES:</b>	CCCCCOC(=O)c1cccc(C(=O)Cl)c1
<b>Mol. weight [g/mol]:</b>	268.74

## Physical Properties

Property code	Value	Unit	Source
gf	-204.99	kJ/mol	Joback Method
hf	-480.35	kJ/mol	Joback Method
hfus	34.25	kJ/mol	Joback Method
hvap	69.98	kJ/mol	Joback Method
log10ws	-4.70		Crippen Method
logp	3.803		Crippen Method
mcvol	205.610	ml/mol	McGowan Method
pc	2115.83	kPa	Joback Method
rinpol	2015.00		NIST Webbook
rinpol	2015.00		NIST Webbook
tb	718.97	K	Joback Method
tc	930.05	K	Joback Method
tf	438.49	K	Joback Method
vc	0.790	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	539.87	J/mol×K	718.97	Joback Method
cpg	553.50	J/mol×K	754.15	Joback Method
cpg	566.23	J/mol×K	789.33	Joback Method
cpg	578.07	J/mol×K	824.51	Joback Method
cpg	589.06	J/mol×K	859.69	Joback Method
cpg	599.22	J/mol×K	894.87	Joback Method
cpg	608.57	J/mol×K	930.05	Joback Method
dvisc	0.0011819	Paxs	438.49	Joback Method

dvisc	0.0006981	Paxs	485.24	Joback Method
dvisc	0.0004523	Paxs	531.98	Joback Method
dvisc	0.0003143	Paxs	578.73	Joback Method
dvisc	0.0002307	Paxs	625.48	Joback Method
dvisc	0.0001767	Paxs	672.22	Joback Method
dvisc	0.0001401	Paxs	718.97	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344716&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344716&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/116-556-6/Isophthalic-acid-monochloride-hexyl-ester.pdf>

Generated by Cheméo on 2024-05-03 16:21:49.895442981 +0000 UTC m=+17042558.816020307.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.