

DL-Alanyl-DL-alanine, N,N'-dimethyl-N'-(but-3-yn-1-yloxycarbonyl)-, but-3-yn-1-yl ester

Inchi: C17H24N2O5
InchiKey: PFGKWNHLVNEQGW-UHFFFAOYSA-N
Formula: C17H24N2O5
SMILES: C#CCCCOC(=O)C(C)N(C)C(=O)C(C)N(C)C(=O)OCCC#C
Mol. weight [g/mol]: 336.38

Physical Properties

Property code	Value	Unit	Source
gf	158.32	kJ/mol	Joback Method
hf	-288.09	kJ/mol	Joback Method
hfus	51.90	kJ/mol	Joback Method
hvap	81.52	kJ/mol	Joback Method
log10ws	-2.37		Crippen Method
logp	0.880		Crippen Method
mvol	269.600	ml/mol	McGowan Method
pc	1755.07	kPa	Joback Method
rinpol	2169.00		NIST Webbook
rinpol	2169.00		NIST Webbook
tb	799.05	K	Joback Method
tc	997.94	K	Joback Method
tf	604.48	K	Joback Method
vc	0.990	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	789.92	J/molxK	799.05	Joback Method
cpg	804.13	J/molxK	832.20	Joback Method
cpg	817.40	J/molxK	865.35	Joback Method
cpg	829.74	J/molxK	898.49	Joback Method
cpg	841.20	J/molxK	931.64	Joback Method
cpg	851.82	J/molxK	964.79	Joback Method
cpg	861.62	J/molxK	997.94	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U392714&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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