

Benzamide, 2-bromo-N-(2-bromobenzoyl)-N-(3-methylbutyl)-

Inchi: InChI=1S/C19H19Br2NO2/c1-13(2)11-12-22(18(23)14-7-3-5-9-16(14)20)19(24)15-8-4-6-

InchiKey: CHZOVBDEBYJQCV-UHFFFAOYSA-N

Formula: C19H19Br2NO2

SMILES: CC(C)CCN(C(=O)c1ccccc1Br)C(=O)c1ccccc1Br

Mol. weight [g/mol]: 453.17

Physical Properties

Property code	Value	Unit	Source
gf	193.80	kJ/mol	Joback Method
hf	-95.62	kJ/mol	Joback Method
hfus	45.54	kJ/mol	Joback Method
hvap	91.78	kJ/mol	Joback Method
log10ws	-7.34		Crippen Method
logp	5.540		Crippen Method
mcvol	279.170	ml/mol	McGowan Method
pc	2183.60	kPa	Joback Method
rinpol	2753.00		NIST Webbook
rinpol	2753.00		NIST Webbook
tb	949.50	K	Joback Method
tc	1199.01	K	Joback Method
tf	618.70	K	Joback Method
vc	1.032	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	776.71	J/molxK	949.50	Joback Method
cpg	788.72	J/molxK	991.08	Joback Method
cpg	799.83	J/molxK	1032.67	Joback Method
cpg	810.19	J/molxK	1074.25	Joback Method
cpg	819.91	J/molxK	1115.84	Joback Method
cpg	829.13	J/molxK	1157.42	Joback Method
cpg	837.98	J/molxK	1199.01	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U407069&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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